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CHAPTER 17 REVIEW Reaction Kinetics MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The reaction for the decomposition of hydrogen peroxide is $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$. List three ways to speed up the rate of decomposition. For each one, briefly

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explain why it is effective, based on collision theory.

17 Reaction Kinetics - David Brearley High School

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Holt Modern Chemistry Review CHAPTER 12 & 13: SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc.

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