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## Chapter 15 Water And Aqueous Systems

**WATER AND ITS PROPERTIES** (pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water.

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**Key Concepts** 15.1 The high surface tension of water and low vapour pressure are due to the hydrogen bonding between the molecules. The structure of ice is a regular open framework of water molecules held together by hydrogen bonds and arranged like a honeycomb. Molecules in liquid water are packed closer together than they are in ice.

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