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nickipolyakov. first chemistry quiz.
Terms in this set (44) matter. anything
that has mass and takes up space.

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chemistry. the study of the composition of matter and the changes that matter undergoes.

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Answer Section. MULTIPLE CHOICE.

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1.3.2. 2. ANS: D DIF: L1 REF: p. 63

OBJ: 3.1.1. 3. ANS: C DIF: L2 REF: p.

63, p. 71 OBJ: 3.1.1. 4. ANS: E DIF:

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ANS: C DIF: L1 REF: p. 66 OBJ: 3.1.2.

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3.1.2. 7. ANS: C DIF: L1 REF: p. 66
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Structure Key Area –Formulae and
reaction quantities. 1. Hydrogen
peroxide is a useful bleaching agent
and is contained in many hair dyes.
Over time, the hair dye becomes less
effective as the hydrogen peroxide

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decomposes forming water and oxygen.

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questions Marks

= 50 kJ mol⁻¹ and ?H = -100 kJ mol⁻¹

B E A = 50 kJ mol⁻¹ and ?H = +100 kJ

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mol^{-1} C E A = 100 kJ mol^{-1} and $\Delta H = +50 \text{ kJ mol}^{-1}$
D E A = 100 kJ mol^{-1}
and $\Delta H = -50 \text{ kJ mol}^{-1}$

CfE Higher Chemistry Unit 1:
Chemical Changes and Structure
In general: Average rate = change in

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variable change in time It is worth mentioning at this stage that the unit of concentration of a solution is measured as moles per litre. In previous courses this may have been shortened to mol/ or even as mol/dm³ At this level the same unit will always be written as mol·l⁻¹.

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o a = 1, b = 3, c = 2 and d = 4 o a = 1,

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b = 3, c = 2 and d = 2 o a = 1, b = 2, c = 2 and d = 2. National 5 Chemistry Revision Page 25 Unit 1 – Chemical Changes & Structures. 6 – The Mole.
a) Gram Formula Mass (GFM) The gram formula mass (GFM) of a substance is known as the mass of one mole.

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