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Chapter 2: Introduction to the Chemistry of Life.
Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things. They form the carbohydrates, nucleic acids, proteins, and lipids (all of which will be defined later in this chapter) that are the fundamental molecular components of all organisms.

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Chapter 2 The Chemistry of Life What are the basic building blocks of all matter? Atoms Describe the structure of an atom protons, electrons, and neutrons. The nucleus (center) of the atom contains the protons (positively charged) and the neutrons (no charge).

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Biology: Chapter 2, The Chemistry of Life. Atom. Nucleus. Electron. Element. An atom is the smallest constituent unit of ordinary matter th.... The nucleus is the small, dense region consisting of protons a.... The electron is a subatomic particle, symbol e^- or β^- , with a....

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A B; atom: the basic unit of matter: nucleus: the center of the atom: electron: a negatively charged particle: element: a pure substance that consists entirely of one type of atom

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Quia - Chapter 2: The Chemistry of Life Vocabulary Review

Chemistry 1405 Chapter 2 1. chemistry. Chemistry is the study of matter, its properties, how and why substances combine or separate to form other substances, and how substances interact with energy. 2. matter. The term matter refers to anything that occupies space and has mass—in other words, the “stuff” that the universe is made of. 3. three different levels of matter – The three ...

1405 - Chapter 2 Vocab(2) (1).rtf - Chemistry 1405 Chapter ...

A compound is a substance formed by the chemical

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combination of two or more elements in definite proportions. (this means that H_2O is water, but H_3O is not). Compounds have different physical and chemical properties from the elements they are made of

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In 1879, William Crooks studied the conduction of electricity through gases at low pressure. He performed the experiment in a discharge tube which is a cylindrical hard glass tube about 60 cm in length. It is sealed at both the ends and ...

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Determination of Molar Mass 2.7 Abnormal Molar
Masses Contents show 1 [...]

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Chapter 2 The Chemistry of Life What do you see when you look at this picture? Is it just a mass of tangled ribbons? Look closely. It's actually a complex pattern of three-dimensional shapes. It represents the structure of a common chemical found inside living cells. The chemical is a protein called kinase.

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Nonmetals; 18.5 Occurrence, Preparation, and Compounds of Hydrogen; 18.6 Occurrence, Preparation, and Properties of Carbonates; 18.7 Occurrence, Preparation, and Properties of Nitrogen

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2. What relationship exists between the mass number of an element and isotopes of that element? 3. Explain the difference between ionic and covalent bonds. 4. Compare and contrast adhesion and cohesion. Below is a guide for Chapter 2 Test

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Chapter 2 chemistry of life. Dehydration synthesis.

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Hydrolysis. Subatomic particles that make up the at....
The atomic number. A chemical reaction in which two
molecules are bonded together.... A chemical process
that splits a molecule by adding water. Protons,
Electrons and Neutrons. The number of protons.

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...

This chapter looks at atoms, bonds, pH and organic
molecules. Good review of chemistry we see in
microbiology.

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Concepts Class Date Section Review 2-4 Completion
On the lines provided, complete the following sentences. 1. Chemical reactions that energy often occur spontaneously. 2. During a chemical reaction, chemical bonds are 3. Biological catalysts, or enzymes, act by lowering the required for a reaction. 4.

Biochemistry 11 Inquiry - Home

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